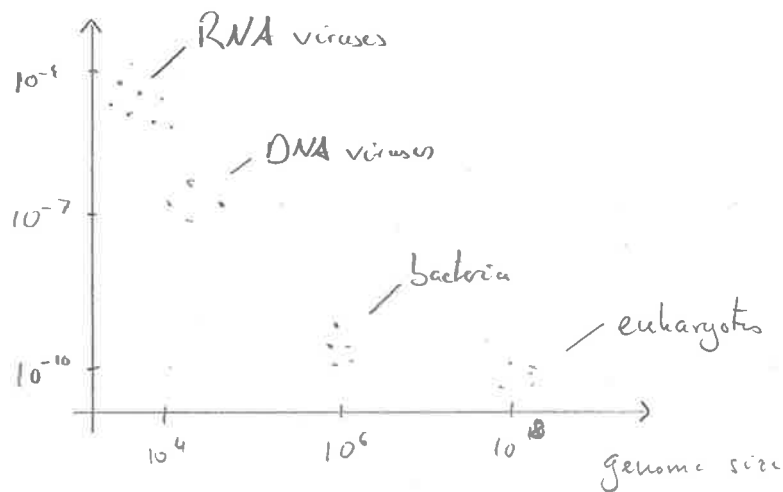


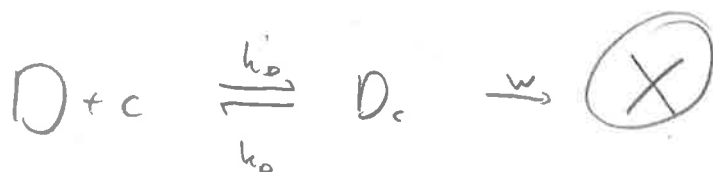
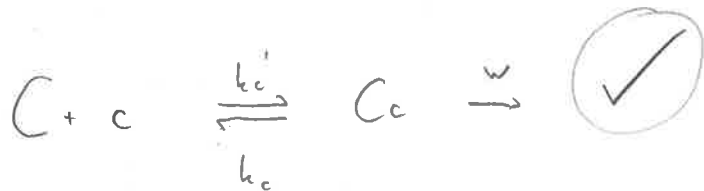
- Observation: biochemical reactions have very high fidelity
- DNA replication: 1 in  $10^{16}$
- how is such accuracy achieved?
- is accuracy costly? how costly?

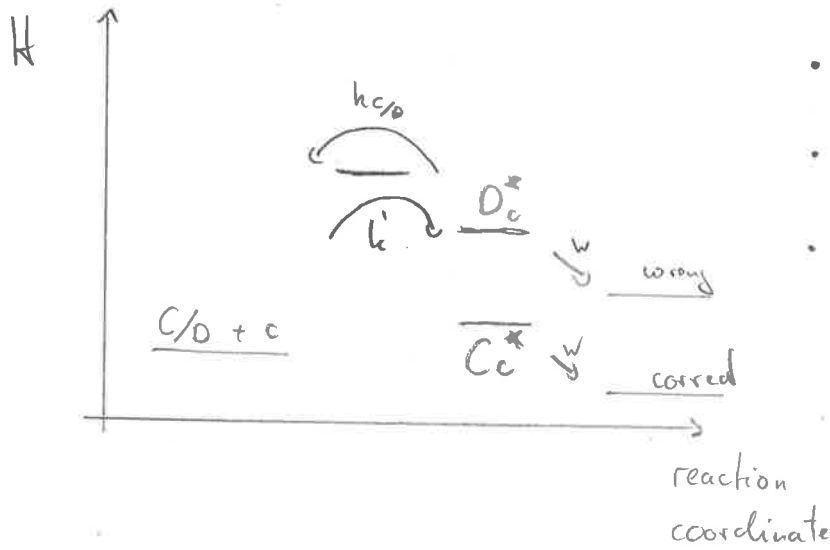


- roughly  $\mu \sim \frac{1}{L}$  (exception eukaryotes bc multiple chromosomes)
- error free replication:  $e^{-\mu L}$   
(needed to avoid Muller's Ratchet & genome degradation)

### Energy of discrimination

- competing reactions of "correct" and "wrong" substrate





- on rates the same
- off rates discriminate
- $k_{c/o} \gg w$

• how much correct/wrong product is formed?

- transition state population is determined by the balance

$$[C_c^*](w + k_c) = [C] k_c \Rightarrow [C_c^*] = \frac{[C] k_c}{w + k_c}$$

- product formation

$$w[C_c^*] = w \frac{[C] k_c}{w + k_c}$$

→ analogously

$$w[C_o^*] = w \frac{[O] k_o}{w + k_o}$$

⇒ ratio

$$\frac{w[C_c^*]}{w[C_o^*]} = \frac{[C] \frac{w + k_o}{w + k_c}}{[O] \frac{w + k_o}{w + k_o}} \rightarrow \frac{[C]}{[O]} \frac{k_o}{k_c} = \frac{[C]}{[O]} e^{-\frac{\Delta G}{RT}}$$

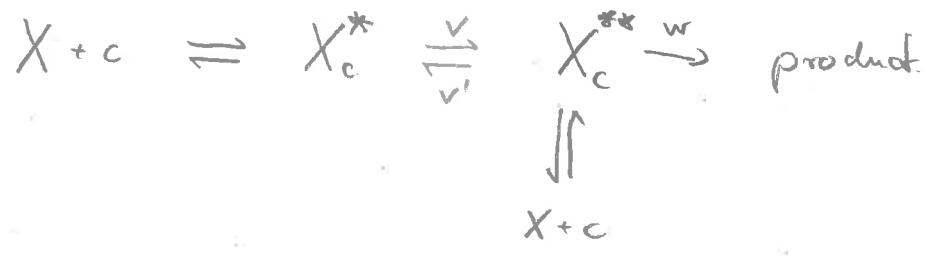
$\Delta G$  = energy of discrimination

⇒ discrimination is bounded by energetic differences of the transition state

• DNA nucleotide pairings  $\sim 2.5kT$   $e^{-\Delta} \approx 0.01 \gg 10^{-5}$

how is higher accuracy achieved?

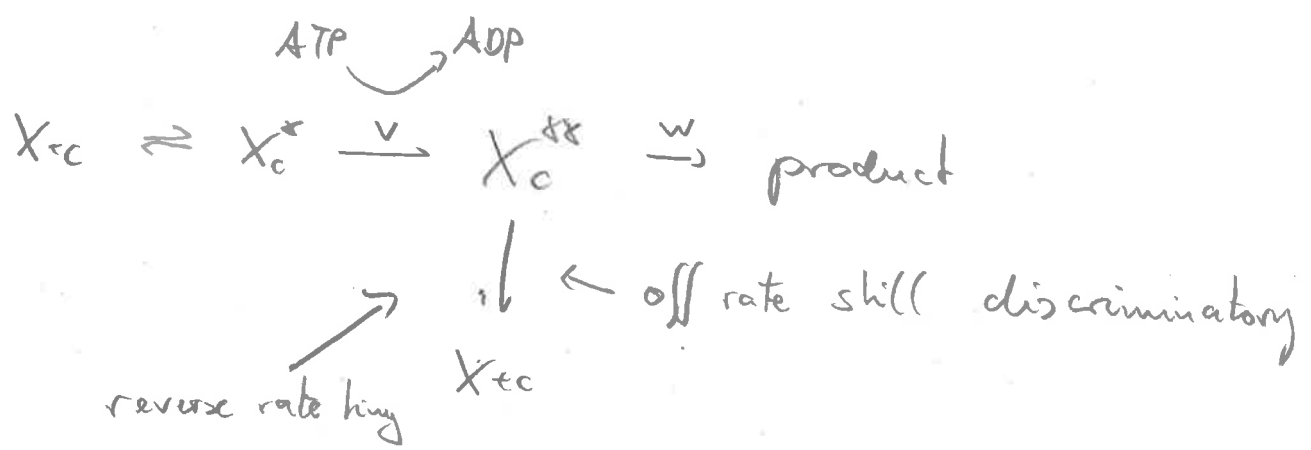
Accuracy requires energy



- detailed balance implies that reversible intermediate steps don't help

- for  $w \ll$  other reactions,  $X_c^{**}$  is at equilibrium  $\Rightarrow e^{-\Delta G/wT}$

- need of an irreversible step

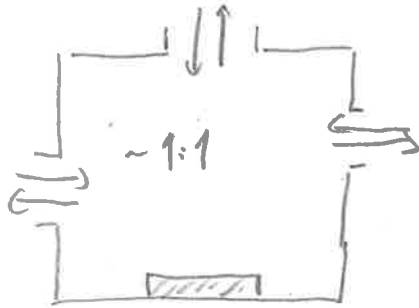


- irreversibility by coupling to hydrolysis

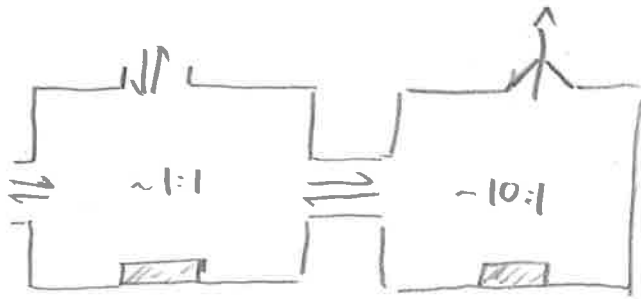
# Silly metaphor

(4)

- consider a museum with an artist that 10% of pple like.
- they spend 10min in these rooms, others 1min



10% stay 10x longer  
roughly 1:1 ratio



- room with exit only results in 10 fold enrichment
- exit only is crucial!

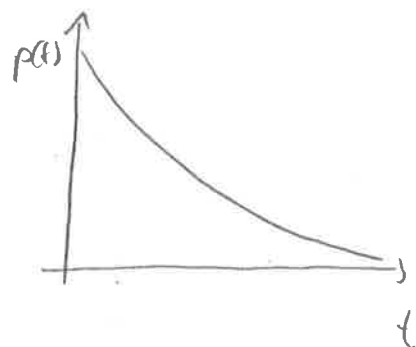
- 
- $X_c^*$  is  $e^{-\Delta G/RT}$  fold enriched
  - adding an additional step  $\rightarrow$  another  $e^{-\Delta G/RT}$  enrichment
  - all dependent on off rates that depend on correct/wrong
  - more steps lead to more accurate reactions
  - typical implementation: conformational transitions of the enzyme or polymerase

## Interpretation as a delay

(5)

- considers a  $X_c^+$  complex
- prob of still being in state  $X_c^+$

$$\frac{dp}{dt} = -(k_x + w)p \rightarrow e^{-(k_x + w)t}$$

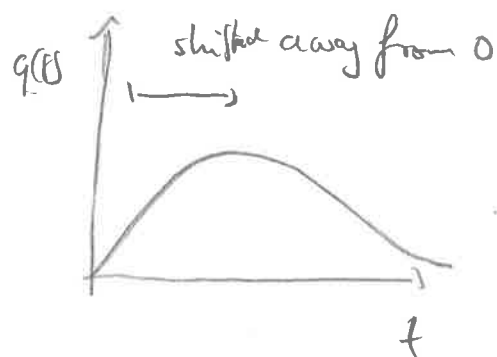


- adding an extra intermediate

$$\frac{dp}{dt} = -(k_x + w)p$$

$$\frac{dq}{dt} = v p - (v + k_x)q$$

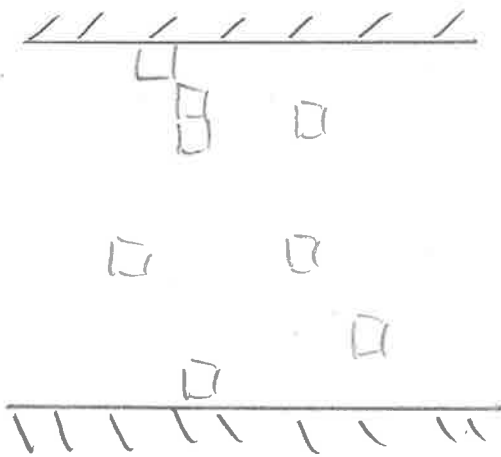
$$q(t) = w e^{-(v+k_x)t} \int_0^t e^{-(w-v)t'} dt'$$



# Phase transitions

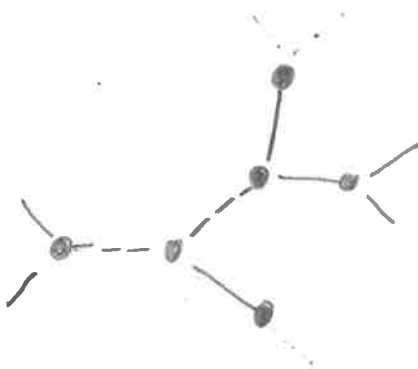
(6)

- Global properties change qualitatively with parameters



- at what filling fraction is there a path from top to bottom?
- when does a liquid turn into a gel?
- ice - water - steam

## Bethe lattice percolation



- prob. of being connected to infinity
- $Q$  = not connected via a particular link

$$Q = (1-p) + p Q^{z-1}$$

$$z=2 \rightarrow p=1 \text{ or } Q=0$$

$$z=3 \rightarrow Q = (1-p) + p Q^2$$

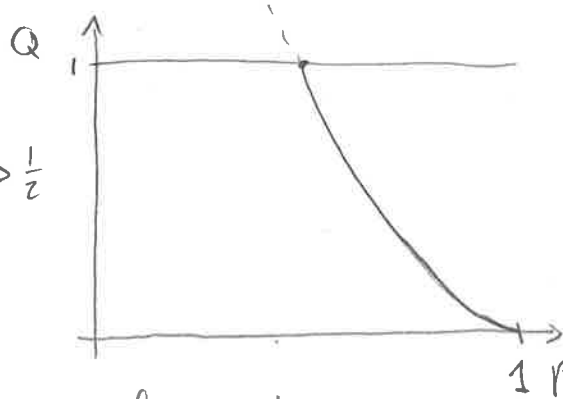
$$Q = \frac{1}{2p} \pm \sqrt{\frac{1}{4p^2} - \frac{1-p}{p}} = \frac{1}{2p} \left( 1 \pm \sqrt{1 - 4p(1-p)} \right) = \frac{1}{2p} (1 - 2p + 1) = \frac{1-p}{p}$$

The probability that any given node is connected

(6)

$$P = p(1 - Q^2)$$

$$= p\left(1 - \frac{(1-p)^2}{p^2}\right) \quad | \quad p > \frac{1}{2}$$



$\Rightarrow$  for  $p > \frac{1}{2}$  there is a finite chance...

- cluster size distribution has long tails at  $p = p_c$  otherwise exponential
- behavior at the transition is universal (independent of details of the lattice etc, but dependent on dimensionality)